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	L 61617-65 ACCESSION NR: AP5015597	!		
•	10 <sup>10</sup> -10 <sup>17</sup> spin/g, which corresponds to one unpaired spin for 10 <sup>11</sup> -10 <sup>7</sup> anthracene molecules. In all samples, light-excited fluorescence and the kinetics of the photoconductivity drop were studied. The relationships obtained indicate that			-
	photoconductivity drop were studied. The relationships changes in the range of paramagnetic center concentrations from 1011- 5 x 10 <sup>12</sup> spin/g.		•	
	ASSOCIATION: Institut khimichaekoy finiki ikademii neuk 8%8 (Institute of Chemical Physics, Academy of Sciences, 888R)		•	-
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Proc	L 8151-66 EWT(m)/EWP(j)/T RM		-
	ACC NR. AP5027690 / SOURCE CODE: UR/0062/65/000/010/1875/1877		
	ACC NR. AP5027690 SOURCE CODE: UR/0062/65/000/010/1875/1877  AUTHOR: Berlin, A. A.; Cherkashin, M. I.; Kisilitsa, P. P. 445	ó	
	ORG: Institute of Chemical Physics, Academy of Sciences SSSR (15)		
	(Institut khimicheskoy fiziki Akademii nauk SSSR)		÷.
	TITLE: Polymerization of beta-iodophenylacetylene 7	<b>12</b>	
	SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 10, 1965, B		
	TOPIC TAGS: polymerization, catalytic polymerization, polymer, linear polymer, polymerization catalyst		
	ABSTRACT: The thermal and catalytic polymerization of beta-iodophenyl- acetylene were investigated to study the effect of different substitue-	G	
٠.	nts on the polymerization of acetylenic compounds. Thermal polymerization at 150 C and catalytic polymerization with triethylaluminum-	•	
	titanium chloride complexes (optimum 70 C, using (C2H5)3Al-TiCl3 with Al:Ti = 1:1) gave polymers which were stable to atmospheric oxidation	_	
	at room temperature but which oxidized at 450-500 C, splitting out icdine and forming three-dimensional structures. Diels-Alder reactions,	6	
	bromination and IR spectral data helped establish that the first stage		
	of this reaction is polymerization at the triple bond to form		
	Card 1/2 UDG: 51/2.952+51/7.362		
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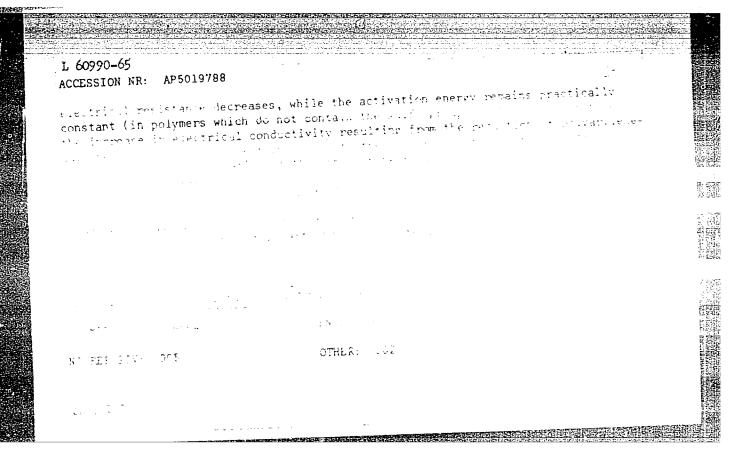
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	ACC NR: AP5025029 44.55 SOURCE CODE: UR/0286/65/000/016/0083/0083	75 - 12 7 - 13 7	
	AUTHORS: Korolev, G. V.; Kondrat'yeva, A. G.; Berlin, A. A. Wiff		
		- 1	!
	ORG: none TITLE: Method for obtaining polymers on the basis of acrylic compounds. Class 39,		
	No. 173941 (		
	SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 16, 1965, 83		
	TOPIC TAGS: polymer, polymerization, acrylic polymer, ascorbic acid, inhibitor,		í
	monomer		÷
	ABSTRACT: This Author Certificate presents a method for obtaining polymers on the basis of acrylic compounds by polymerizing corresponding monomers or oligomers containing a quinone type inhibitor. To increase the rate and depth of polymerization, a reducing agent, e.g., ascorbic acid, is introduced into the reaction mixture.		
を受ける。 のは、 のは、 のは、 のは、 のは、 のは、 のは、 のは、	ABSTRACT: This Author Certificate presents a method for obtaining polymers on the basis of acrylic compounds by polymerizing corresponding monomers or oligomers conbasis of acrylic compounds by polymerizing corresponding monomers or oligomers conbasis of acrylic compounds by polymerizing corresponding monomers or oligomers con-		
のはないのは、ならればない。 一般のない こうない ないしゅうしゅう かんしょう しょうしゅうしゅう	ABSTRACT: This Author Certificate presents a method for obtaining polymers on the basis of acrylic compounds by polymerizing corresponding monomers or oligomers containing a quinone type inhibitor. To increase the rate and depth of polymerization, a reducing agent, e.g., ascorbic acid, is introduced into the reaction mixture.		
	ABSTRACT: This Author Certificate presents a method for obtaining polymers on the basis of acrylic compounds by polymerizing corresponding monomers or oligomers containing a quinone type inhibitor. To increase the rate and depth of polymerization, a reducing agent, e.g., ascorbic acid, is introduced into the reaction mixture.		
	ABSTRACT: This Author Certificate presents a method for obtaining polymers on the basis of acrylic compounds by polymerizing corresponding monomers or oligomers containing a quinone type inhibitor. To increase the rate and depth of polymerization, a reducing agent, e.g., ascorbic acid, is introduced into the reaction mixture.		

	I 61:01-56 EWT(1)/EPA(s)-2/EWT(m)/EPF(c)/ETC/EWG(m)/EWP(1)/T/EWA(h)/ETC(m) YJP(c)		—
,	ACC NR. AP5028544 DS/WI/AT/RM SOURCE CODE: UR/0286/65/000/020/0161/0161		
	INVENTOR: Berlin, A. A.; Dubinskaya, A. M.		· · · · ·
	ORG: none TITLE: Preparation of heat-resistant polymers. Class 39, No. 151811		+
	SOURCE. Byulleten' izobreteniy i tovarnykh znakov, no. 20, 1965, 161		
	TOPIC TAGS: heat resistant polymer, semiconducting polymer, ion exchange polymer		
Sad.		1.5	. 6
	ABSTRACT: An Author Certificate has been issued for a preparative method for heat- resistant polymers with ion exchange and semiconductor properties by The method in-	1.25	
	resistant polymers with ton exchange and second substituted quinones with carboxylic acid amides volves polycondensation of halogen substituted quinones with carboxylic acid amides such as acetamide, formamide, or dimethylformamide.		
	SUB CODE: GC, MT/ SUBM DATE: 280ct61/ ATD PRESS: 4/39		
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	: Method of Stabilizing bolvo	rganosiloxanes	against therma	l-oxidation agi	ing.
Class	39, No. 176067 announced by to V. Lomonosov (Moskovskiy inst	the Moscow Inst	titute of Fine	Chemical Techno	
	44,55			•	
SOURC	E: Byulleten' izobreteniy i to	ovarnykh znako	v, no. 21, 1965	, 48	
TOPIC	TAGS: polysiloxane, stabilize	er, oxidation	inhibition		
	ACT: An Author Certificate has	s been issued	for a method of	stabilizing po	
	osiloxanes to prevent thermal-	oxidative agin			
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L 1145-66 EVT(m)/EVP(j)/T RM UR/0076/65/039/009/2281/2284 ACCESSION NR: AP5023692

541.124/.12

TITLE: Hechanism of inhibition of oxidative processes by compounds with conjugated

systems /

SOURCE: Zhurnal fizicheskoy khimii, v. 39, no. 9, 1955, 2281-2284

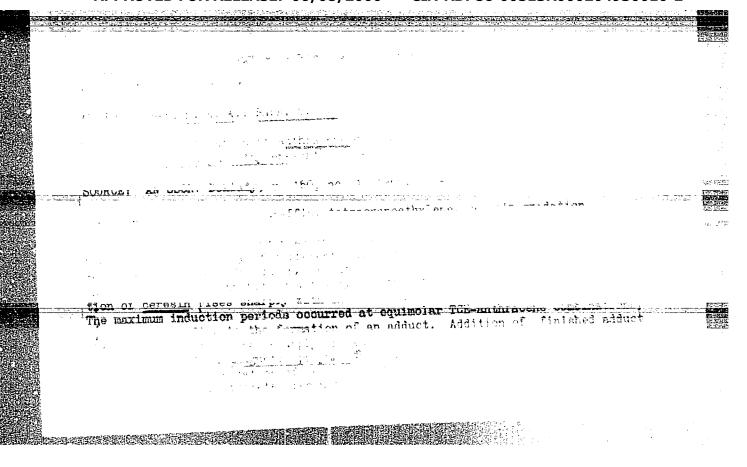
AUTHOR: Bass, S. I.; Berlin, A.

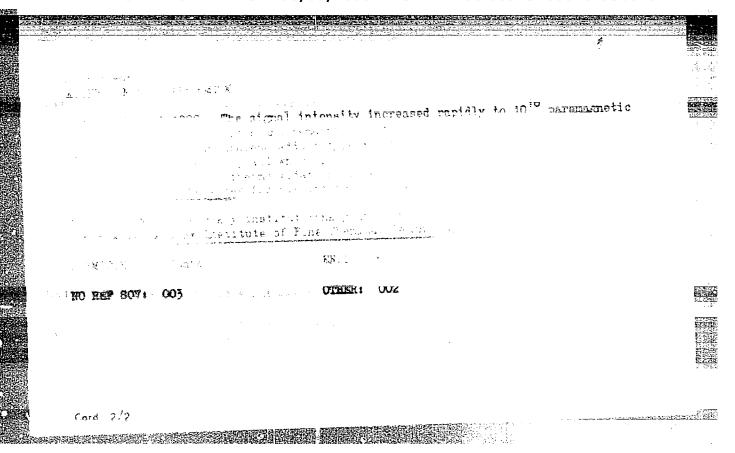
TOPIC TAGS: oxidation inhibition, anthracene, paramagnetic material

ABSTRACT: In order to determine the mechanism governing the activating influence of the addition of paramagnetic particles on anthracene; the consumption of the latter during oxidation of ceresin was studied with and without the addition of a paramagnetic fraction isolated from products of thermal treatment of anthracene and containing  $2 \times 10^{18}$  paramagnetic particles per gram of substance. It was found that the mechanism of the inhibiting action of anthracene differs substantially from thatwhich usually occurs during inhibition of oxidizing processes by known antioxodants: no appreciable loss of anthracene is observed during the induction period, and the duration of the latter is increased. It is postulated that the paramagnetic centers-

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ACCESSION N	R: AP5023692					6	
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ASSOCIATION	: Hoskovskiy	institut to	nkoy khimi	cheskoy tek	UDOTOĞIT (HOE	CON TUBLY.	-
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ENT(m)/ENP(j) T. 13525-66 SOURCE CODE: UR/0190/65/007/012/2057/206 ACC NR: AP6001861 AUTHORS: Berlin, A. A.; Aseyeva, R. M.; Aseyev, Yu. G. ORG: Institute of Chemical Physics AN SSSR (Institut khimicheskoy fiziki AN SSSR) TITLE: Effect of halogen in a conjugated chain upon the reactivity of polyvinylene. 61st report in the series Conjugated Polymers 14 55 SOURCE: Vysokomolekulyarnyye soyedineniya, v. 7, no. 12, 1965, 2057-2062 TOPIC TAGS: polyolefin, hydrocarbon, polyvinyl chloride, polymer chemistry, hydrogenation, halogenation, maleic anhydride / IKS 14 spectrophotometer ABSTRACT: Catalytic hydrogenation, halogenation, addition of maleic anhydride and molecular hydrogen to polyvinylene (I) and to polyvinylene chloride [II] were studied. These addition reactions were of interest as it was observed that the presence of chlorine in the conjugated structure of polyvinylene affects the mechanism of formation and properties of the carbon skeleton of the macromolecules derived by thermal treatment. I and II we're prepared by dehydrochlorinating polyvinyland and polyvinylidene chloride, using sodium amylate at equimolar ratios, as described by A. A. Berlin, R. M. Aseyeva, C. I. Kalyayev, and e. L. Frankevich (Dokl. AN SSSR, 114, 1042, 1962); and R. M. Aseyeva, Yu. C. aseyev, A. A. Berlin, and V. I. Restochtin (Zh. strukt. khimil, 6, 47, 1965). Hwdry genstion as performed in decalin with Mi-Al catalyst, at 1000 for I and at 200 atm of H2. 000) 678.01:54+678.742 Cord 1/2

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SHMURAK, I.L.; UZINA, R.V.; BERLIN, A.A.

Some factors determining the formation of chemical bonds on the boundary of adhesive - substrate separation. Kauch. i res. 24 (MIRA 18:10) no.9:23-26 165.

1. Nauchno-issledovatel'skiy institut shinnoy promyshlennosti i Institut khimicheskoy fiziki AN SSSR.

BOOUSLAVSKIY, L.I.; SHERLE, A.I.; BERLIN, A.A.

Electrophysical properties of films of polymeric complexes of tetracyanoethylene with a metal. Zhur. fiz. khim. 38 no.5:1118-1125 My '64. (MIRA 18:12)

1. Institut elektrokhimii AN SSSR i Institut khimicheskoy fiziki AN SSSR. Submitted March 18, 1963.

BASS, S.I.; BERLIN, A.L.

Meanwrism of the inhibition of oxidizing processes by compounds with a conjugated system. Zhur. fiz. khim. 39 nc.9: 2281-2284 S 65. (MIRA 18:10)

1. Meskovskiy institut tenkoy khimleheskoy tekhnologit.

RM EWP(j)/EWT(m)/T IJP(c) 30779-66 UR/0080/65/038/012/2804/2807 SOURCE CODE: AP6022139 ACC NRI Berlin, A. A.; Shau-ch'uan, Huang-Shau-Tsyuan', Khuan; Maloshitskiy, A. S. AUTHOR: ORG: none TITIE: Synthesis and investigation of the products of heptane and dodecane chlorophosphorylation SOURCE: Zhurnal prikladnoy khimii, v. 38, no. 12, 1965, 2804-2807 TOPIC TAGS: chlorinated organic compound, alkylphosphine, phosphorylation, hydrocarbon, oxidation, polycondensation ABSTRACT: The effect of the PCl3:RH ratio on production of dichlorohydrides of alkylmono- and alkyldiphosphinic acids was studied. The investigation was conducted with heptane and dodecane. The data obtained shows that with an increase in the PCl3:RH ratio, the yield of alkyldiphosphinic acid chloride rises, and the yield of the undistilled residuo also rises. This consists of more highly phosphorylated paraffins or their reaction products. The optimal PCl3:RH ratio can be selected from a graph of the yield of phos. phorylation products as a function of the PCl3:Ri ratio. It must be noted that at low PCl3:RH ratios the rate at which the reaction mixture becomes colored rises sharply. This is evidently related to exidation of the hydro-Card 1/2 0056

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ACC NR. AP5027182 SOURCE CODE: UR/0076/65/039/010/2571/2573

AUTHOR: Bass, S. I.; Berlin, A. A.

ORG: Moscow Institute of Fine Chemical Technology im. M. V. Lomonosov (Moskovskiy Institut tonkoy khimicheskiy tekhnologii)

TITLE: Synergistic effects in the inhibition of oxidation processes

SOURCE: Zhurnel fizicheskoy khimii, v. 39, no. 10, 1965, 2571-2573

TOPIC TAGS: oxidation inhibition, synergy, hydrocarbon, anthracene, phosphite, sulfide

ABSTRACT: A study has been made of the oxidation inhibition in such paraffins as hexadecane or ceresin by synergistic mixtures in which one component is a hydroperoxide reducing agent and the other is a free-radical acceptor. The experiments were conducted with mixtures of phosphites or sulfides with anthracene or anthracene heat treatment products. In phosphite mixtures, the phosphite acts as the reducing agent and anthracene acts as the free-radical acceptor. Phosphites are themselves free-radical acceptors and anthracene frees them for their reducing function. Experiments showed the effectiveness of the synergistic mixtures. This effectiveness can be improved not only by selection of the proper reducing agent, but also by varying the length of reconjugated chain and the paramagnetic particle concentration of the free-radical acceptor. The authors express their gratitude to Academician S. S. Medvedev

**Cord** 1/2

L 5102-66 EWT(m)/EPF(c)/EW7(1)

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L 13812-66 EWT(m)/EWP(1)/T RM

ACC NR, AP6002470

SOURCE CODE: UR/0191/66/000/001/0003/0006

AUTHORS: Berlin, A. A.; Kefeli, T. Ya.; Belkin, A. A.; Ragimov, A. V.; Liogon'kiv, B. I.; Brikenshteyn, Kh. A.

ORG: none

TITLE: On the catalytic and inhibiting influence of certain polysulfophenylquinones on the telomerization condensation reaction

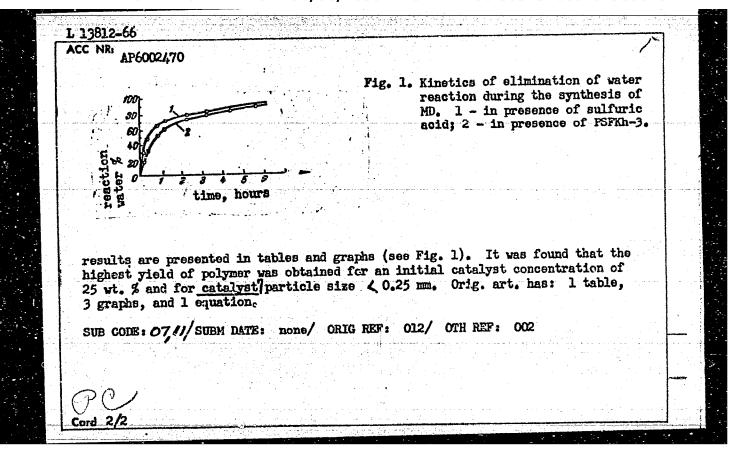
SOURCE: Plasticheskiye massy, no. 1, 1966, 3-6

TOFIC TAGS: polymer, polymerization, catalytic polymerization, high polymer, polymerization kinetics, polymerization rate

ABSTRACT: The catalytic and inhibiting effects of polysulfophenylquinone (obtained by reacting p-benzoquinone with bis-diazotized benzidine-disulfo-2,21 acid (PSFKh-3) in the mole ratio of 1:3) on the condensation telomerization/of digo-esteracrylates, dimethylmethacrylate of diethylene glycol (MD) and dimethylmethacrylate-bis-diethyleneglycol)phthalate (MDF-1) were studied. The method used was that described by A. A. Berlin, T. Ya. Kefell, and G. V. Korolev (Khim. promass No. 12, 12, 1962). The kinetics of water elimination during the synthesis of MDF in the presence of sulfuric acid and PSFKh-3, and the effect on PSFKh-3 of the ion exchange properties of synthesized esters were investigated. The experimental

Card 1/2

UDC: 678.764.43:678.044.1:547.567



BERLIN, A.A.; PARINI, V.P. [deceased]; AL'MANBETOV, K.

Activation of cis-trans-isomerization of dimethyl maleinate by polymers with a conjugate system. Dokl. AN SSSK 166 no.3:595-597 Ja 166. (MIRA 19:1)

1. Institut khimicheskoy fiziki AN SSSR. Submitted June 13, 1965.

L 21252-66 EWT(m)/EWP(j)/1/ETC(m)-6 WW/WR UP (037)/66/000/001/0003/0006	
ACC NR: AP6008397 (A) SOURCE CODE: GR/03(4/00/000)	
AUTHOR: Babich, V. F.; Sivergin, Yu. M.; Berlin, A. A.; Rabinovich, A. L.	
ORC: Institute of Chemical Physics AN SSSR, Moscow (Institut khimicheskoy 1121k1	
TITLE: Correlation between the equilibrium modulus of high elasticity and the number of cross-links in rigid network structure polymers	
SOURCE: Mekhanika polimerov, no. 1, 1966, 3-6	,
TOPIC TAGS: crosslinking, polymer structure, elastic modulus, temperature dependence, temperature effect, equilibrium	
ABSTRACT: The dependence of the equilibrium modulus of the high elasticity of polymers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates of olygoesteracrylates on temperature was investigated. The modulus was shown to mers of olygoesteracrylates ol	
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	<u>L 39714-66</u> EWP(j)/EWT(m)/T IJP(c) RM/GD-2	r.	
•	ACC NRI AF6007962 (A) SOURCE CODE: UR/0191/66/000/003/0005/0007		
	AUTHOR: Berlin, A. A.; Kaplunov, I. Ya.; Barminov, V. A.		
•	CRG: none		
	TITLE: Compatibility of polyvinyl chloride with oligo-ester-acrylates and some properties of their products		
:	SOURCE: Plasticheskiye massy, no. 3, 1966, 5-7		
	TOPIC TAGS: polyvinyl chloride, acrylic plastic, polymerization, oligomer, solid physical property, electric property		
	ABSTRACT: A study was made to obtain the optimal formulation of the polymer- oligomer systems (clathrate polymers) having the best physicochemical proporties. Using		
	different weight ratios of the reagents, polyvinyl chloride (PVC) and a polymerizable olimpier were polymerized at 1250 for 2 hr. The compatibility of the reagents was		
	determined by a nephelometric study of films of the polymers produced. Linear oligo- mers TGM-3, MGF-9, Fand MBF-1, and short-chain organosilicon/DEMFS-2m were mixable		
	with PVC at a wide range of concentrations. Branched oligomers 7-1 and 7-20 could be mixed at low concentration. Clear films were obtained with 30-40% 7-1 or 7-20. The		
Ş	polymers obtained hardened at 140-1450 and had a higher liquidus temperature than PVC (due to the cross-linked bonds formed). Their thermal stability was 40-450 higher		
	and their electrical properties were better than those of PVC. When extracted with		
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L 37209-66 EWT(m)/EWP(j) ACC NR: AP6014412 RM SOURCE CODE: UR/0062/66/000/004/0746/0747 AUTHOR: Berlin, A. A.; Gafurov, Kh. M.; Mayorov, N. S.; ORG: Institute of Chemical Physics. Academy of Sciences SSSR (Institut TITLE: Effect of local activation in zone melting of polynuclear aromatic SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 4, 1966, 746-747 TOPIC TAGS: polynuclear hydrocarbon, anthracene, chemical purity, paramagnetic material, zone melting ABSTRACT: The possibility of using zone melting to completely purify condensed aromatic hydrocarbons of paramagnetic particles (PP) was investigated. Synthetic anthracene still contained 10% 10% PP per gram after zone melting and PP were found in samples which had no detectable PP before zone melting. Similar observations were made with pyrene and hence zone melting will not free polynuclear aromatic hydrocarbons of PP. Orig. art. has: 1 table and 1 figure. SUB CODE: 07/ Card 1/1 mc/ SUBM DATE: 16Aug65/ ORIG REF: 005

APPROVED FOR RELEASE: 06/08/2000 CIA-RDP86-00513R000204930010-2"

UDC: 538.113

L 44180-66 Sir(a)/aur(j)/r ACC NRI AP6011233 (A)SOURCE CODE: UR/0413/66/000/006/0074/0074 INVENTOR: Berlin, A. A.; Ostroumova, L. Ye.; Gusev, M. N.; Martsenitsena, ORG: none TITLE: Method of obtaining oligomers which can be polymerized. Class 39, SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 6, 1966, 74 TOPIC TAGS: oligomer, polymerization, methacrylic acid ABSTRACT: An Author Certificate has been issued for a method of obtaining oligomers which can be polymerized. A chloroparaffin or a dehydrochloroparaffin is subjected. to interaction at 120-140C with a potassium or sodium salt of methacrylic acid in a polar initiating solvent, such as a demethylformamide. [Translation] SUB CODE: 07 SUBM DATE: 18Aug64/ aught Card 1/1 UDC: 678. 6

IJP(c) NW/RM SOURCE CODE: UR/0191/66/000/007/0008/0009 AUTHOR: Korolev, G. V.; Kondrat'yeva, A. G.; Borlin, A. A. ORG: none TITLE: Chemical regulation of inhibitor activity in the radical-chain polymerization of monomers and oligomers SOURCE: Plasticheskiye massy, no. 7, 1965, 8-9 TOPIC TAGS: radical polymerization, chain reaction polymerization, iodine, ascorbic acid, hydroquinone, polymerization kinetics ABSTRACT: The paper describes some methods for chemically regulating the activity of inhibitors of radical-chain polymerization by introducing into the polymerization system suitable admixtures which increase or decrease the effectiveness of quinone-type inhibitors. The effect of such admixtures on hydroquinene and benzoquinene was determined from the change in the polymerization kinetics of methacrylates /(methyl methacrylate, polyester acrylates). Polyester acrylate of brand 7-20 was used in the experiments. The polymerization was carried out at 50°C in the presence of the initiator trila (10) Toding construction of the initiator trilater trilat trile (AD). Iodine admixtures were found to increase the effectiveness of quinonetype inhibitors considerably, whereas ascorbic acid admixtures deactivate the inhibitors almost completely. By treating I2 as an oxidant and ascorbic acid as a reductant C TDC: 578.045

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RM/WW L 32169-66 EWP(j)/EWI(m)/T SOURCE CODE: UR/0413/66/000/007/ ACC NR. AP6012139 INVENTOR: Berlin, A. A.; Kefeli, T. Ya.; Pilippovskaya, Yu. M.; Korolev, V. V.; Marhonina, L. I.; Leogon kiv. B. ORG: none TITLE: Preparation of polyacrylate esters. Class 39, No. 180335 SOURCE: Izobreteniya, promyshlennyye obrastsy, tovarnyye snaki, no. 7, 1966, 57 TOPIC TAGS: polyester, acrylate, polymerisation ABSTRACT: An Author Certificate has been issued describing a method of preparing polyacrylate esters by low-temperature polymerisation in bulk of monomeric and oligomeric acrylate esters in the presence of peroxide initiators. To speed up the process the system benzene peroxide plus polyazophenylene plus filler with a developed surface such as PK-3, K-40/is suggested as the initiator. The polymerisation is carried out in the presence of an inhibitor of medium potency, for instance benzoquinone or diphenylamine. SUB CODE: 11,07 SUBM DATE: 22Aug62 UDC: 678.674'2'0 Card 1/1 he

L 46994-66 EWP(1)/EWT(m)/T IJP(c) RM/WW SOURCE CODE: UR/0191/66/000/008/0018/0021 ACC NRI AP6027275

AUTHOR: Borlin, A. A.; Ignatyuk, A. G.; Kefeli, T. Ya.; Sel'skaya, O. G.; Sivergin,

Yu. M.; Komleva, L. K.

ORG: none

TITIE: Xylitcl oligoester acrylates/and some properties of their polymers

SOURCE: Plasticheskiye massy, no. 8, 1966, 18-21

TOPIC TAGS: acrylate, xylitol, polycondensation, adipic acid, sebacic acid, phthalic anhydride

ABSTRACT: The synthesis and polymerization of oligoester acrylates (OEA) based on xylitol and some properties of products of their curing were studied. The synthesis was carried out by the condensation telemerization method and involved the reaction of xylitol with adipic acid, sebacic acid or phthalic anhydride, with methacrylic acid as the monofunctional telogen, H2SO4 or p-toluensulfonic acid as the catalyst and hydroquinone as the inhibitor. As indicated by the amount of water formed by the reaction and by the analysis of physicochemical properties of the synthesized OEA, the polyesterification reaction in toluene does not involve xylitol itself, but its 1,4-monoanhydride (xylitan). The degree of dehydration of xylitol depends on the nature of the catalyst: it was much greater in the presence of H2SQ, than in the presence of p-toluenesulfonic acid. The conditions of synthesis of the product of the reaction with

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UDC: 678.674'65'52'28.01:539.2

L 46994-66

ACC NR: AP6027275

phthalic anhydride were studied most thoroughly. The amount of methacrylic acid was found to have a marked effect on the formation of this oligoester and its separation from the reaction mixture. A study of the physicomechanical properties of the three cured oligoesters showed that as the flexibility of the oligomer block of the original oligoester increases, the specific impact strength of the polymers rises, and the hardness and bending strength fall off. The oligoesters were found to have a satisfactory thermal stability and resistance to thermal-oxidative degradation. Orig. art. has: 4 figures and 2 tables.

SUB CODE: 11/ SUBM DATE: none/ ORIG REF: 011/ OTH REF: 002

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L 44592-66 EWT(m)/EWP(j)/T IJP(c)ACC NR: AP6013275 SOURCE CODE: UR/0413/66/000/008/0078/0078 INVENTOR: Berlin, A. A.; Berkman, Ya. P.; Shuter, L. M. ORG: none TITLE: Method of obtaining graft copolymers of carboxymethylcellulose and unsaturated monomers. Class 39, No. 180791 SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 8, 1966, 78 TOPIC TAGS: copolymer, monomer, copolymerization, polymerization initiator, graft copolymer ABSTRACT: An Author Certificate has been issued for a method of obtaining graft copolymers of carboxymethylcellulose and unsaturated monomers in the presence of initiators of graft copolymerization reaction in a hydrogen medium. To obtain water-insoluble, film-forming products, carboxymethylcellulose is subjected to preliminary treatment with watersoluble peroxide compounds. The treatment of carboxymethylcellulose by UDC: 678.546.11.9-416:678.744.325 Card 1/2

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L 44578-66 EWT(m)/EWP(j)/T IJP(c) WW/RM AP6015660 (A) SOURCE CODE: UR/0413/66/000/009/0073/0073 ACC NR: INVENTOR: Frunze, N. K.; Berlin, A. A.; Braynes, M. Ya.; Shaydurova, N. K. ORG: none TITLE: Method of obtaining compositions suitable for photopolymerization. Class 39, No. 181280 \5 SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 9, 1966, 73 TOPIC TAGS: photopolymerization, copolymer, polymerization initiator, photosensitivity, oligoether ABSTRACT: An Author Certificate has been issued for a method of obtaining compositions suitable for photopolymerization using the acrylic copolymer series, an oligorhythmic compound, a polymerization initiator, and a sensitizer. To expand the variety of photosensitive compositions, a butylmethacrylate copolymer with methacrylamide is used as the copolymer, and an oligoether such as dimethacrylate-Card 1/2 UDC: 771. 531. 678. 744. 32-134. 548. 3:66. 095. 265

ACC NR. AP6015660 bis-diethyleneglycoltion]	phthalate is suggested as the olygorhyth	mic compound.	O [Transla- [NT]
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EWT(m)/EWP(j)/T IJP(c) RM ACC NR: AP6015679 (A) SOURCE CODE: UR/0413/66/000/009/0078/0078 INVENTOR: Korolev, G. V.; Smirnov, B. R.; Yarkina, V. V.; Berlin, A. A. ORG: none TITLE: Preparation of formulations which can be polymerized when exposed to light. Class 39, No. 181300 SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 9, 1966, 78 TOPIC TAGS: photopolymerization, polymerization ABSTRACT: This Author Certificate introduces a method of preparing formulations suitable for photopolymerization. The formulations contain a polymer base, a compound that can be polymerized and a polymerization initiator. To stabilize the material and to plasticize it temporarily during processing, the compound containing in its chain groups. is suggested as suitable for polymerization. [LD] [Translation] SUB CODE: 11/ SUBM DATE: 25Feb65/ Card 1/1 POM UDC: 771, 531, 678, 745, 6:66, 095, 265

L 09253-67 EWT(m)/EWP(j) JJP(c) RM/WW ACC NR. AP6029910 SOURCE CODE: UR/Oh13/66/000/015/0086/0087 INVENTORS: Bass, S. I.; Borlin, A. A.; 58 Yarkina, V. V.; Sbinar, L. A. OilG: none TITLE: A method for imparting heat resistance to hardened phenolaldehydride resins, Class 39, No. 184431 SOURCE: Izobret prom obraz tov zn, no. 15, 1966, 86-87 TOPIC TAGS: thermal stability, thermal process, resin, heat resistant plastic ABSTRACT: This Author Certificate presents a method for imparting heat resistance to hardened phenolaldehydride resins. This is accomplished by adding to them (prior to their hardening) stabilizing compounds capable of interlinking and containing 1016\_\_ 1019 paramagnetic particles per gram. To produce high-temperature stabilization (at temperatures on the order of 400C), polyphenyl acetylene pr hardened phenolaldehydride resins (heat treated at 300--5000 in an atmosphere of an inert gas or in a vacuum) are used as stabilizers. SUB CODE: 11/ SUBM DATE: 13Feb65 Cord 1/1 UDC: 678.048.9:678.632

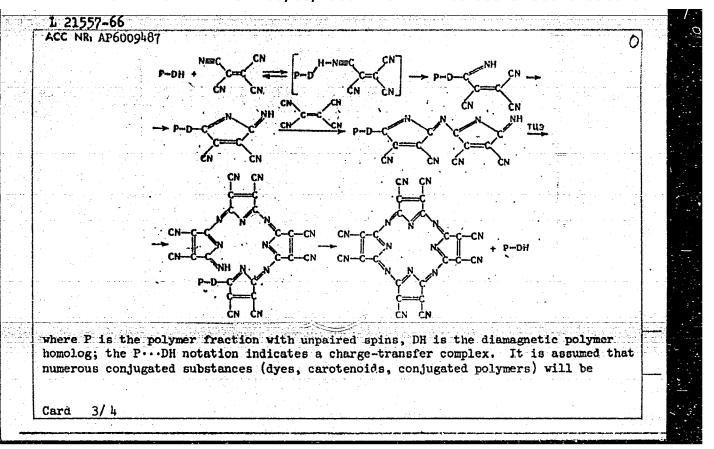
LJP(c) EWT(m)/EWP(j)/T L 32756-66 SOURCE CODE: UR/0190/66/008/004/0699/0702 ACC NR: AP6012714 (o / AUTHOR: Afonskiy, V. K.; Berlin, A. A.; Yanovskiy, D. M. ORG: Institute of Organochlorine Products and Acrylates (Institut khloroganicheskikh produktov i akrilatov) TITLE: Effect of anthracene compounds obtained by thermolysis on thermal and photooxidative degradation of polyvinyl chloride SOURCE: Vysokomolekulyarnyye soyedineniya, v. 8, no. 4, 1966, 699-702 TOPIC TAGS: polyvinyl chloride, polymer, paramagnetic material, pyrolysis, high temperature effect, anthracene ABSTRACT: A study was made of the effect of the products of anthracene compounds obtained by pyrolysis on thermal and photo-oxidative degradation of polyvinyl chloride. The addition of anthracene treated at 450C has a stabilizing role in polyvinyl chloride degradation. The dependence of the amount of hydrogen chloride liberated during polymer degradation on the additive concentration and on the quantity of paramagnetic particles is of extremal nature. The stabilization effect is decreased with the temperature. The relation between the inhibiting behavior of the anthracene pyrolysis products and the energy of singlet-triplet transition was established. Orig. art. has: 2 figures and 1 table. [Based on authors' abstract.] 11, 07/ SUBM DATE: 28Apr65/ ORIG REF: 010/ OTH REF: 001/ SUB CODE: UDC: 678.01:54+678.743 Card 

L 46149-66 EWT(m)/EWP(j)/T IJP(c) WW/RM ACC NR AP6031946 · SOURCE CODE: UR/0080/66/039/009/2035/2038 AUTHOR: Al'shits, I. M.; Anikina, T. A.; Berlin, A. A.; Grad, N. M.; Levitskaya O. M.; Mudrov, O. A.; Pagasyan, S. A.; Tsubina, Kh. V. ORG: none TITLE: A new oligomeric binder for glass-reinforced flashes SOURCE: Zhurnal prikladnoy khimii, v. 39, no. 9, 1966, 2035-2038 TOPIC TAGS: glass reinforced plastic, binder, resin MA-3, triethylene glycol dimethacrylate, TGM-3, polyethylene-glycol-maleate-phthalate, MS-1 ABSTRACT: A new binder for glass-reinforced plastics has been prepared from triethylene glycol dimethacrylate (TGM-3) in which the content of the stabilizer - hydroquinone - was decreased to 0.04% instead of the conventional 0.03 to 0.20%, and from polyethylene glycol malcate phthalate (MS-1 resin) by heating the components to 80C and a vigorous stirring. This mixture was prepared in MS-1:TGM-3 ratios of 2:3 and 1:1; the products had viscosities of 50 and 150 centipoises at 20C respectively, which offers an advantage as compared with the viscosity of 250-430 centipoises of MA-3| resin (specifications: VTU 30-12044-61 of the ISNKh) which is used for manufacturing glass-reinforced plastics in the USSR. The mechanical and technological properties of this new binder make possible its use for impregnating glass fabrics and for applying the method of contact molding. The time of gel formation of the new Card 1/2 UDC: 678

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4 CL22/-00 EWT(m)/EWP(1)/T.
ACC NR: AP6009467 RM/WW SOURCE CODE: UR/0020/66/167/001/0091/0094 AUTHOR: Berlin, A. A.; Matveyeva, N. G. ORG: Institute of Chemical Physics, Academy of Sciences SSSR (Institut khimicheskoy TITLE: Polymarization and copolymerization of tetracyanocthylene under the effect of polymers with a conjugated system SOURCE: AN SSSR. Doklady, v. 167, no. 1, 1966, 91-94 TOPIC TAGS: organic semiconductor, semiconducting polymer, quasiradical polymeriza-ABSTRACT: The feasibility has been studied of using paramagnetic conjugated polymers as catalysts in the "quasi-radical" polymerization and copolymerization of low reactivity monomers such as tetracyanoethylene (I), anthracene, naphthacene, and pentacene. I was polymerized alone or copolymerized with anthracene, naphthacene, or pentacene. Soluble fractions of polyphenylene or polyanthryl were used as the catalyst in various concentrations so as to vary the unpaired spin concentration. Polymerization was carried out at 2000 and 10-3 mm Hg. It was found that the polymer or copolymer yield increased with spin concentration. In the case of the copolymerization of I with anthracene, the highest yield was obtained at a I/anthracene ratio of 4/1. The copolymers were black infusible powders insoluble in the common organic Cord 1/4

he copolymers onlo olymers was 5-6 nd had an activat henylene also con	polymers were more 3500 the homopolyme by went as high as orders of magnitude tion energy of 8—1 polymerized with I. arge-transfer compl	30-35%. The ele as high as the kcal/mol. Na	ed fully, while lectrical conducat of anthracence phthacene, penta	weight losses in ctivity of the co- e, i.e., 10 <sup>-9</sup> nho/co acene, and poly-	1,



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	21423-66 ENT(m)/EWP(i)/T/ETC(m)-6 VM/RM SOURCE CODE: UR/0190/66/008/003/0540/9547-2/	
I.	ACC NR. AP6010120 (A) SOURCE CODE: DR/0190/00/	
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	AUTHOR: Beylin, A. A.: Ragimov, A. V.; Liop )'kiy, B. I.; Belova, G. V.	
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	ORG: Institute of Chemical Physics, AN SSSR (Institut khimicheskoy fiziki AN SSSR)	43.2
	investigation of polyarylenequinones	
	3. 1966, 540-541	
	TOPIC TAGS: organic semiconductor, semiconducting polymer, heat resistant polymer,	
	morte TAGS: organic semiconductor, semiconducting polymer,	
	ABSTRACT: New polyarylenequinones exhibiting redox and ion-exchange properties of particles and ion-exchange properties and ion-exchange prope	
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3	acid (IV). The polymer, 355.	
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where M is -CH=CH- or absent. Structures A were obtained in the case of equimolar benzoquinone/amine ratios, and structures AB, in the case of excess amine. The polymers were light- to dark-brown powders. I and II were partly soluble in acetone, dimethylformsmide, pyridine, nitrobenzene, quinoline, conc. H<sub>2</sub>SO<sub>4</sub>, and aqueous alkalies Sulfonic acids III and IV were soluble or swelled in water, ethyl alcohol, dimethylformsmide, and pyridine. Number-average molecular weights of the soluble fractions of I and II were 900—1500. It was shown that the reaction of benzoquinone with salts of aromatic or bisdiazo (as well as diazo) leads to the reduction of part of Cord 2/3

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L 23049-66 EWT(m)/EWP(j) WW/RM ACC NR AP6012708 SOURCE CODE: UR/0190/66/008/004/0627/0634 AUTHOR: Cherksshine, L. G.; Berlin, A. A. Institute of Chemical Physics, AN SSSR (Institut: khimicheskoy fiziki AN SSSR) TITLE: Synthesis and investigation of certain physicochemical: properties of polymeric phthalocyanines Vysokomolekulyarnyye soyadineniya, v. 8, no. 4, 1966, 627-634 TOPIC TAGS: organic semiconductor, semiconducting polymer, polyphthalo-ABSTRACT: Poly(magnesium phthalocyanine) (I) and its metal-free analog have been synthesized from 1,2,4,5-tetracyanobenzene. 7 Polymer I was prepared by heating 1,2,4,5-tetracyanobenzene, urea, and powdered magnesium under argon at 300C, followed by appropriate purification. The metal-free homolog was prepared from I by treatment with concentrated H2SO4. For comparison, poly(magnesium phthalocyanine) was also prepared from pyromellitic acid, which was identical to I in composition. Elemental-analysis, solubility, and IR spectroscopy data suggested that I exists in the form of linear, and planar parquet-like structures. Electrical measurements were conducted with powder samples **Card** 1/2

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L 24295-66 ENT(1)/ENT(m)/ENP(j)/T ACC NR. AP6009797 IJP(c) SOURCE CODE: UR/0062/66/000/002/0331/0332 AUTHOR: Berlin, A. A.; Gafurov, Kh. M.; Parini, V. P. (Decessed) ORG: Institute of Chemical Physics, Academy of Sciences, SSSR (Institut khimicheskoy fiziki Akademii nauk SSSR) TITLE: The effect of peramagnetic particles on the crystallization temperature of materials with a conjugated system? SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 2, 1966, 331-332 TOPIC TAGS: anthracene, phenanthrene, conjugate bond system, crystallization, supercooling, paramagnetism, optic property, photoelectric property, chemical purity ABSTRACT: The effect of paramagnetic particles on crystallization and other properties of anthracene and phenanthrene was studied. Samples of the W -complex of anthracene containing paramagnetic particle concentrations of 10 % - 10 % spin/gm were prepared by thermal treatment of the compound. There is little dependence of crystallization temperature on paramagnetic particle concentration in the 10 spin/gm range. There is general correlation between changes in the Cord 1/2 UDC: 547.672+542.65

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ACC NR: AP6017886

SOURCE CODE: UR/0062/66/000/005/0945/0945

AUTHOR: Berlin, A. A.; Liogon'kiy, B. I.; Shamrayev, C. M.; Belova, G. V. BO

ORG: Institute of Chemical Physics, Academy of Sciences SSSR (Institut khimicheskoy fiziki Akademii nauk SSSR)

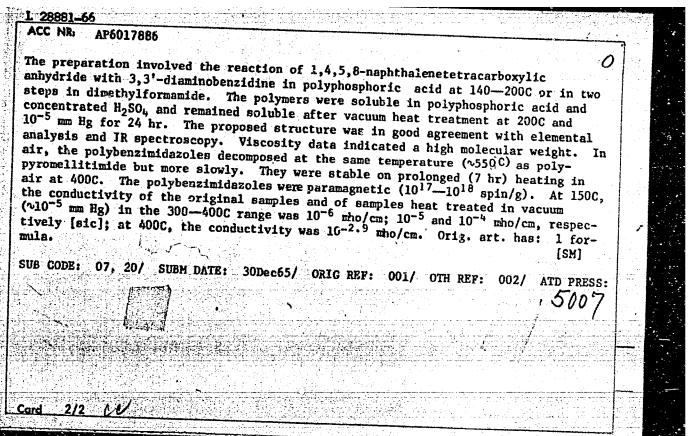
TITLE: New high-thermal-stability polymers with semiconducting properties: Bource: AN SSSR. Izvestiya Seriya khimicheskaya, no. 5, 1966, 945

TOPIC TACS: organic semiconductor, semiconducting polymer, heat resistant polymer, polybenzimidazole

ABSTRACT: New high-thermal-stability polybenzimidazoles — poly[naphthoylenebis—levated temperatures:

October 1/2

DOC: 542,91-541,64-541,67



L 23273-66 EWT(m)/EWP(3)/T/ETC(m)-6 LJP(c) WW/RM ACC NR. AP6012721 SOURCE CODE: UR/0190 UR/0190/66/008/004/0736/0743 AUTHOR: Berlin, A. A.; Matveyeva, N. G. ORG: Institute of Chemical Physics AN SSSR (Institut khimicheskoy TITLE: Tetracyanoethylene polymerization and copolymerization catalyzed by paramagnetic polymers SOURCE: Vysokomolekulyarnyye soyedineniya, v. 8, no. 4, 1966, 736-743 TOPIC TAGS: organic semiconductor, semiconducting polymer, polytetracyanonitrile, quasiradical polymerization ABSTRACT: The feasibility has been studied of using paramagnetic conjugated polymers as catalysts in the "quasi-radical" polymerization of tetracyanoethylene (I) and its copolymerization with certain other conjugated compounds. I was homopolymerized or copolymerized with anthracene, naphthacene, pentacene, or polyphenylene. Soluble fractions of polyphenylene or polyanthryl were used as the initiator in various concentrations. It was found that in the presence of conjugated polymers I is capable of homo- and copolymerization. The product yield increased with initiator concentration. thracene was 4/1 in the appropriate copolymer. The I-anthracene The molar ratio I/an-UDC: 66.095.26+678.13+678.745

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ACC NR: AP7004065

SOURCE CODE: UR/0190/67/009/001/0045/0051

AUTHOR: Berlin, A.A.; Cherkashin, M.I.; Kisilitsa, P.P.; Kushnerev, M.Ya.

ORG: Institute of Chemical Physics, AN SSSR (Institut khimicheskoy fiziki AN SSSR)

TITLE: Study of structural changes in electrical and physical properties of polyphenylacetylene in the course of heat treatment

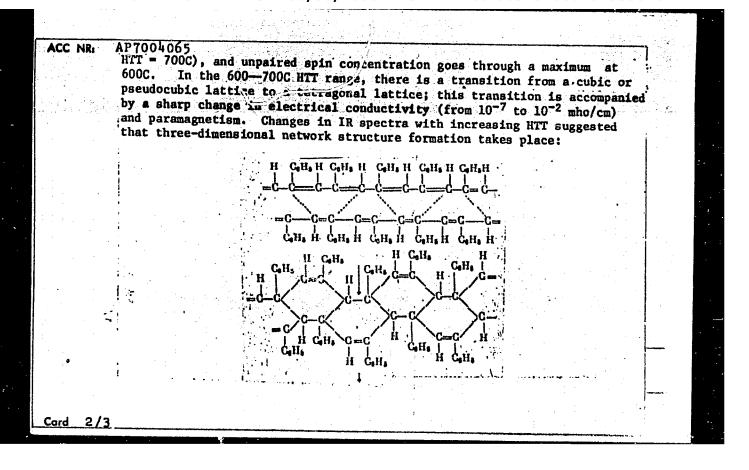
SOURCE: Vysokomolekulyarnyye zoyedineniya, v. 9, no. 1, 1967, 45-51

TOPIC TAGS: pyrolysis, polymer heat effect, polymer structure, electric property, crystallography, phenyl compound, acetylene, conjugated polymer

ABSTRACT: A study has been made of the effect of heat treatment at 300—700C in an inert medium on the morphology, chemical structure, electrical properties, and paramagnetic properties of polyphenylacetylene. The electrical measurements were carried out for pressed pellet specimens at 20—400C. It was shown that heat treatment causes substantial changes in electrical, paramagnetic, and crystallographic properties. As the heat treatment temperature (HTT) increases from 330 to 700C, crystallinity and conductivity increase (from 10<sup>-15</sup> to 10<sup>-2</sup> mh./cm), activation energy for conduction decreases (from 1.50 ev at HTT = 400C to 0.19 ev at

Card 1/3

UDC: 678.01:53/54+678.76

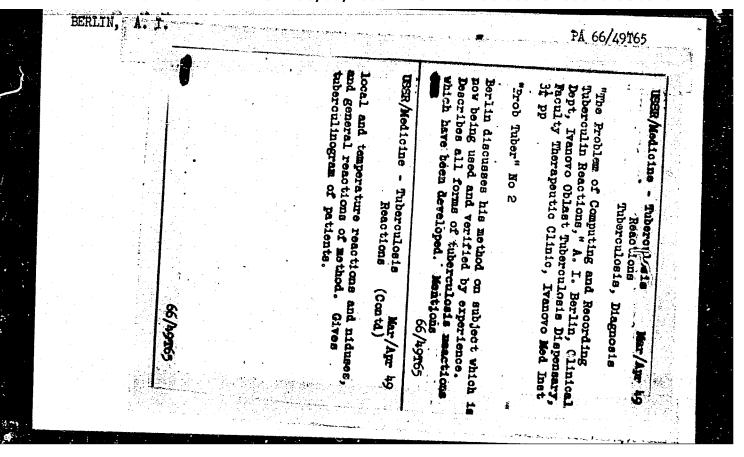


ACC NRI AP7004065 Gelfa II Celfa To determine the effect that the three-dimensional network formation and crystallinity win polyphenylacetylene have on conductivity, the properties of the phenylacetylene-p-diethylbenzene copolymer and the polyphenylacetylene-p-diethylbenzene block copolymer which has a three-dimensional network structure were studied. The crystalline structure was crystalline for the copolymer and amorphous for the block copolymer but both had conductivity of the order of only  $10^{-16}\,$  mho/cm. This indicates that crystallinity and a three-dimensional network structure are not sufficient conditions for a high conductivity in conjugated polymers. SUB CODE: 11, 20/ SUBM DATE: 300ct65/ ORIG REF: ATD PRÉSS: 5114 004/ OTH RFF: 003

BERLIN, A. I. Dr. Med. Sci.

Dissertation: "The Therapeutic Application of Bromine in Tuberculosis During the Flash Period." Central Inst. for Advanced Training of Physicians. 18 Nov 47.

SO: Venternyaya Molkva, Nov, 1947 (Project #17836)



BERLIN, AI.

#### HERLIN A. I.

Opyt terapii obostrenii tuberkulesa na osnove uchemiia shkoly I. P. Pavlova o vysshei nervani deiatel'nosti. /Therapy of acute tuberculosis based on Pavlov's theory of the physiology of the nervous system/ Probl. tuberk., Moskva No. 5 Sept-Oct 50 p. 15-22.

1. Of the Tuberculosis Clinic attached to the Faculty Therapeutic Clinic of Ivanovo Medical Institute (Head — Prof. L. I. Vilenskiy) and of the Oblast Tuberculosis Dispensary (Head Physician — Candidate Medical Sciences G. M. Dubitskiy).

BERLIN, A.I., prof.; PETRASH, L.G.

Acid-release function of the stomach during PMS therapy in tuberculosis. Probl. tub. 35 no.6:57-90 157. (NIRA 12:1)

1. Iz fakul tetskoy terapevticheskoy kliniki Ivanovskogo meditsinskogo instituta i Oblastnogo protivotuberkuleznogo dispansera.

(PARA AMINOSALICYLIC ACID, eff.

gastric acidity in tuberc. ther. (Rus))

gastric, eff. of PAS in tuberc. (Rus))

33922

5/079/62/032/002/007/011 D243/D303

5.3630

Fel'dman, I.Kh. and Berlin, A.I.

TITLE:

AUTHORS:

Synthesis of stereoisomeric cyclic phosphorus organic

compounds

PERIODICAL: Zhurnal obshchey khimii, v. 32, no. 2, 1962, 575-579

TEXT: In the present investigation all the theoretically possible 3dimensional isomers of the cyclic, phosphamide esters of 3-4-dimethyl-5-phenyl - 2-N'-bis- $(\hat{\beta}$ -chloroethyl)-amine-2,1,3 -phosphoxazolidine were obtained by condensation of N-bis (/3-chloroethyl) dichlorophosphoxamide and optically active  $\underline{d}$  - and  $\underline{l}$ - ephedrine. Cyclic phosphanide stereochemistry, especially the difference in physiological activity between dextro and laevo forms of asymmetric phosphorus organic compounds, iz stated to be of great interest for treating certain forms of cancer. The condensation of d-pseudoephedrine and N-bis (eta -chlorethyl) dichlorephosphoxamide: a solution of 36.9g. of d-pseudoephedrine in 600 ml. of dry benzene was added, drop by drop, with stirring, over 40-50 minutes,

33922 S/079/62/032/002/007/011 D243/D303

Synthesis of stereoisomeric ...

to a solution of 19.3g. N-bis-( $/^2$ -chloroethyl)-dichlorophosphoxamide in 100 ml. of dry benzene, the reaction temperature not exceeding 30°C. After stirring for 4 hours, the d-pseudoephedrine chlorhydrate precipitate (29.4g) was filtered off, the filtrate vacuum-evaporated and the residue recrystallized frequently for 1-2 days. A solution of the latter in absolute alcohol was diluted with dry ether until it clouded, boiled for 3-5 minutes with wood charcoal and kept in a refrigerator for 24 hours. After filtration and vacuum evaporation the residue was redissolved in benzene and, on adding a small quantity of ether, product Ia Eq.(2) separated out, followed by a mixture of the latter and its diastereoisomer, Ib Eq.(2).

Card 2/5

33922

S/079/62/032/002/007/011 D243/D303

Synthesis of stereoisomeric ...

After fractional crystallization from ether 7.9g. of Ia and 7.5g of Ib were obtained. The condensation of d-ephedrine and N-bis-(\$\beta\$-chloroethyl)-dichlorophosphoxamide: a solution of 32.5g. of d-ephedrine in 50 ml. of absolute ether was added, while stirring, over 2 hours to a solution of 17g. N-bis-(\$\beta\$-chlorethyl)-dichlorophosphoxamide in 200 ml. absolute either. After being kept for 2 hours at 18 C, stirring was continued for 16 hours while the solution boiled. The ether solution was separated from the d-ephedrine chlorhydrate formed, boiled for 10 minutes with wood charcoal, filtered, evaporated to 200-250 ml. and left for 2-3 days in a refrigerator, when crystals of substance IIIa separated Eq.(4).

The mother liquor was diluted with absolute ether and the resultant clear

Card 3/5

33922

Synthesis of stereoisomeric ...

S/079/62/032/002/007/011 D243/D303

solution poured off and evaporated to 50-60 ml. After standing for 10-15 days crystals of IIIa formed again. This cycle was repeated until crystals of IIIb appeared. After frequent crystallization from benzene-ether mixture and from other 6g. of IIIa and 2.8g. IIb respectively were obtained. The reactions with \*-pseudoephedrine, d.-pseudoephedrine, b-ephedrine and d. t-ephedrine were carried out in a similar way. Details of the products are given. The authors state that the substances obtained are the first examples of optically active phosphorus organic compounds in which there are two asymmetric carbon atoms and an asymmetric phosphorus atom. There are 1 table and 31 non-Soviet-bloc references. The 4 most recent references to the English-language publications read as follows: H.S. Aaron, J. Braun and Th.N.Shryne, J.Am.Chem.Soc., 82, 57 (1960); G.M. Campbell and I.K.Way, J.Chem.Soc., 1960, 5034; M. Green and R.F. Hudson, J. Chem. Soc., 1960, 3129; O.N. Friedman, Proc.Am. Ass. Canc. res. 3, 112, (1960), re. 86.

ASSOCIATION: Leningradskiy khimiko-farmatsevticheskiy institut

Card 4/5

Synthesis of stereoisomeric ...

33922 S/079/62/032/002/007/011

D243/D303

(Leningrad Chemical and Pharmacological Institute)

SUBMITTED:

January 5, 1961

Card 5/5

# FEL'DMAN, I.Kh.; HERLIN, A.I.

Synthesis of steroisomeric cyclic organophospherus compounds. Part 2: Cis- and trans-isomeric derivatives of cyclically bound phosphorus. Zhur.ob.khim. 32 no.5:1604-1607 My 162.

1. Leningradskiy khimiko-farmatsevticheskiy institut.
(Phosphorus organic compounds) (Stereochemistry)

# FEL'DMAN, I.Kh.; HERLIN, A.I.

Synthesis of cyclic stereoisomeric organophosphorus compounds. Part 3: Stereospecificity of the reaction of cyclic phosphoramide ester formation. Zhur.ob.khim. 32 no.10:3379-3381 0 62. (MIRA 15:11)

1. Leningradskiy khimiko-farmatsevticheskiy institut.
(Phosphoric triamide)

BERLIN, A.L.

RT-793 [Significance of the thermoprecipitin reaction in the praxis of examining plague epizootics] K voprosu o znachenii reaktsii termopretsipitatsii v praktike obsledovaniia chumnykh epizootii.

Vestnik Mikrobiologii, Epidemiologii i Paraziologii, 9(1): 54-59, 1930.

1. 12335-63 EPR/EWP(j)/EPF(c)/EWT(m)/BDS AFFTC/ASD PB-4/Pc-4/Pr-4 RP/HW ACCESSION NR: AP3000753 S/0020/63/150/003/0580/0583 73

AUTHOR: Dudina, L. A.; Berlin, Al. Al.; Karmilova, L. V.; Yenikolopyan, N. S.

TIME: Changing the molecular weight by oxidative destruction of polyformaldehyde

SOURCE: AN SSSR. Doklady, v. 150, no. 3, 1963, 580-583

TOPIC TAGS: oxidative destruction, polyformaldehyde

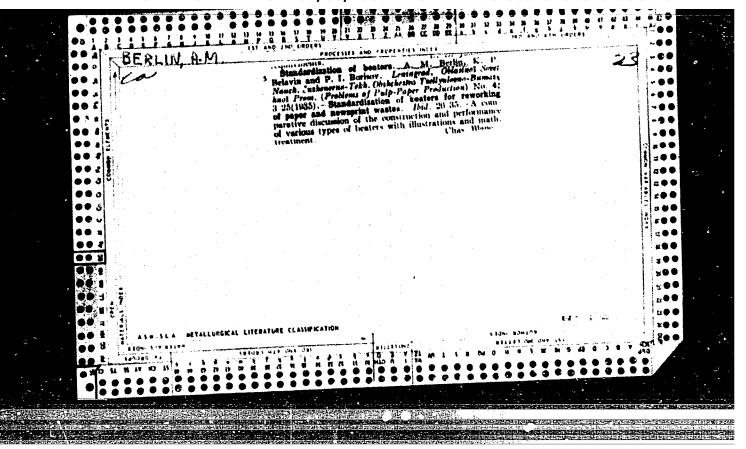
ABSTRACT: In continuation of earlier work by L. A. Dudina, L. V. Karmilova, and N. S. Yenikolopyan (DAW. 150, no. 2, 1963), a study was made to determine whether the 0-initiated decomposition of polyformaldehyde does indeed go according to the equation shown in the enclosure. Experimentally a sharp drop in molecular weight was found, such that its dependence on the degree of conversion is representative of the formation of two stable fragments upon decomposition of the molecule. These fragments could be formed either because the 0 inhibits decomposition of the active center in addition to initiating destruction of the polymer or because of acidolysis of the polymer by the acid from the oxidation process. Calculations also indicated that the above-proposed reaction does not account for the drop in molecular weight. Orig. art. has: 1 figure and 28 equations.

Cord 1/3, Inch of Chemical Physics

BERLIN, Al.Al.; BAFKALOV, I.M.; GOL'DANSKIY, V.I.; YENIKOLOPYAN, N.S.

Kinetic of solid-phase polymerization. Dokl. AN SSSR 160 no.5: 1104-1107 F '65. (MIRA 18:2)

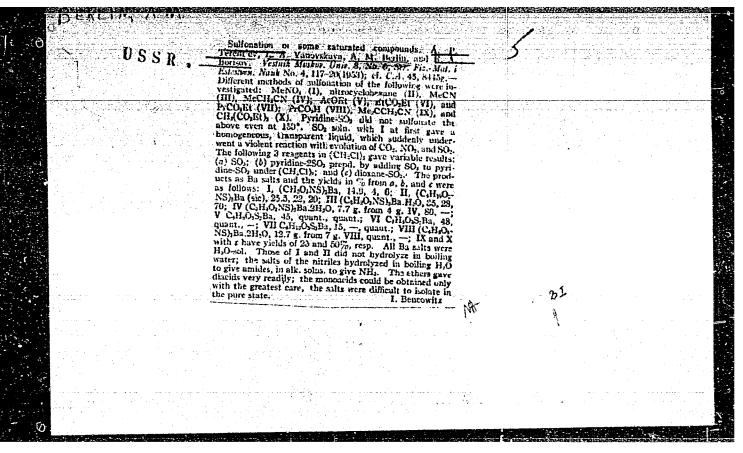
1. Institut khimicheskoy fiziki AN SSSR. 2. Chlen-korrespondent AN SSSR (for Gol'danskiy).

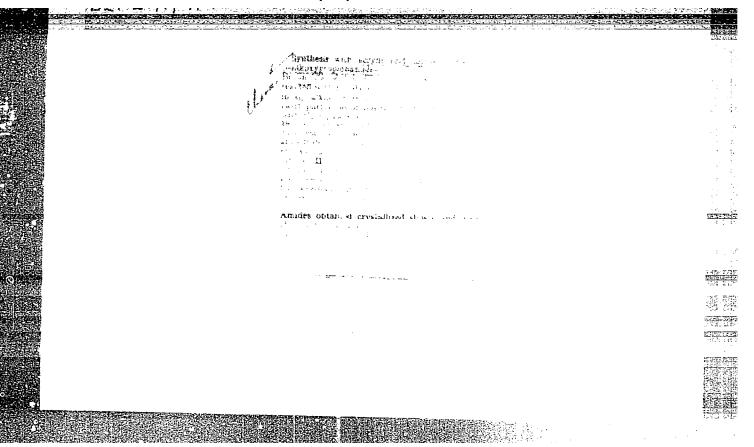


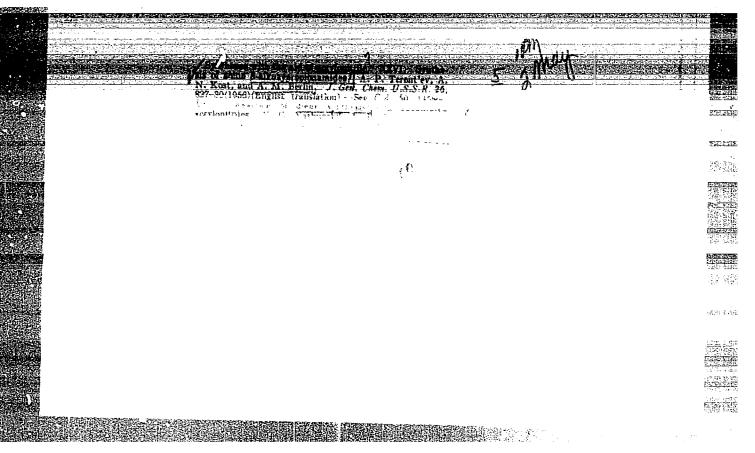
TERRENT' YEV, A.P.; KOST, A.N.; HERLIN, A.M.

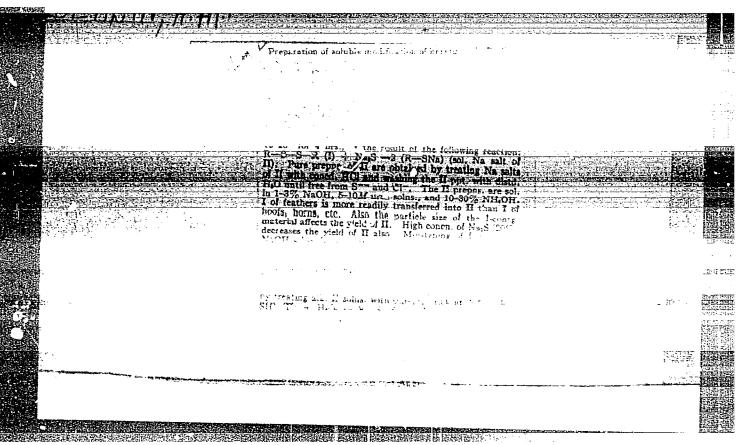
Syntheses with the aid of acrylic acid nitrile. Part 22. New method for the synthesis of pyrrolines. Ehur. ob.khim. 25 no.8:1613-1616
Ag 155. (MIRA 9:2)

1. Moskovskiy gosudarstvennyy universitet. (Pyrroline) (Mitriles)









DERLIN

82101

5.3700

S/062/60/000/07/04/007 B015/B054

**AUTHORS:** 

Nesmeyanov, A. N., Nogina, O. V., Berlin, A. M.,

Kudryavtsev, Yu. P.

TITLE:

Chemical Transformations of Dialkoxy Titanium Oxides

PERIODICAL:

Izvestiya Akademii nauk SSSR. Otdeleniye khimicheskikh nauk,

1960, No. 7, pp. 1206-1214

TEXT: The authors describe the chemical properties of dialkoxy titanium oxides. They continued the investigation of deposition to the>Ti = 0 bond, studied the etherification reactions, and found the substitution of alkoxyls by halogens as well as a substitution reaction of the oxygen bound to titanium with two chlorine atoms. They obtained the first compounds of a hitherto unknown series of titanium-containing organic compounds, i.e., ethoxy-, n-propoxy-, and n-butoxy titanium oxide chlorides. By the action of chlorine on di-n-propoxy- and disobutoxy titanium oxides, they produced the compounds Cl2TiC.2n-C3H7OH and Cl2TiO.2i-C4H9OH. On the reaction of dialkyl dichloro silanes with dialkoxy titanium oxides, the following exchange of oxygen with two chlorine atoms takes place: Card 1/2

Chemical Transformations of Dialkoxy Titanium Oxides

 $(n-c_3H_70)_2$ Ti0 +  $cl_2SiR_2 \longrightarrow (n-c_3H_70)_2$ Ti $cl_2$  + OSiR<sub>2</sub> x. A similar reaction takes place by the action of tetra-n-propoxy silane on the compound  $\text{cl}_2\text{Ti0-2n-c}_3\text{H}_7\text{OH}, \text{ namely } \text{cl}_2\text{Ti0-2n-c}_3\text{H}_7\text{OH} + (\text{c}_3\text{H}_7\text{O})_4\text{Si} \rightarrow \text{cl}_2\text{Ti}(\text{oc}_3\text{H}_7)_2\text{"n-}$  $C_3H_7OH + n-C_3H_7OH + \left|OSi(OC_3H_7-n)_2\right|_x$ . By the linkage of dimethyl di-npropoxy silane with di-n-propoxy titanium oxide, the following compounds were obtained:

There are 8 references: 7 Soviet and 1 British.

ASSOCIATION:

Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR (Institute of Elemental-organic Compounds of the

Academy of Sciences, USSR)

SUBMITTED:

January 23, 1959

Card 2/2

K

S/020/60/134/003/013/020 B016/B054

5.3700 AUTHORS:

Nesmeyanov, A. N., Academician, Nogina, O. V., and Berlin,

A. M.

TITLE: Interaction of Cyclopentadienyl Sodium With Alkoxy Titanium Chlorides

PERIODICAL: Doklady Akademii nauk SSSR, 1960, Vol. 134, No. 3,

pp. 607-608

TEXT: The authors investigated the reactions of cyclopentadienyl sodium with alkoxy titanium chlorides, and isolated mono-T-cyclopentadienyl derivatives of titanium: cyclopentadienyl-triethoxy- and tri-n-propoxy titanium (see Diagram). They are colorless liquids distillable in vacuo and very sensitive to air moisture. The authors indicate the established and calculated yields and molecular weights of the two derivatives. As opposed to the ordinary alkoxy derivatives of titanium, the T-cyclo-pentadienyl-trialkoxy derivatives are not associated in solutions, not even at a concentration of 1.7 moles. The molecular weight of C5H5Ti(OC2H5)3, cryoscopically determined in benzene solution, corresponds

Card 1/3

Interaction of Cyclopentadienyl Sodium With Alkoxy Titanium Chlorides

S/020/60/134/003/013/020 B016/B054

to that of the monomeric compound. The authors proved the structure of  $\pi$ -cyclopentadienyl-triethoxy- and tri-n-propoxy titanium by conversion into the known (Refs. 7,8) cyclopentadienyl-titanium trichloride (see Diagram). Further, the authors allowed ethoxy-titanium trichloride to react with cyclopentadienyl sodium, and determined and calculated the physical constants of  $(C_5H_5)_2\mathrm{Ti}(OC_2H_5)\mathrm{Cl}$ . The  $\pi$ -cyclopentadienyl-

trialkoxy derivatives of titanium are decomposed by alcohols at 70-80°C within 1 h, while corresponding tetraalkoxy titanium and cyclopentadiene are formed. The latter was isolated as cyclopentadienyl thallium. By reactions of \$\pi\$-cyclopentadienyl-tri-n-propexy titanium with acetyl chloride, the authors obtained mixed chloride alcoholates of \$\pi\$-cyclopentadienyl titanium: 1) cyclopentadienyl-di-n-propoxy-titanium chloride which is disproportionated by vacuum distillation; 2) cyclopentadienyl-n-propoxy-titanium dichloride. Both compounds are greenish-yellow, viscous liquids decomposing under the action of air moisture, but still storable at lower temperatures. There are 8 references: 2 Soviet and 2 US.

ıX

Card 2/3

Interaction of Cyclopentadienyl Sodium With Alkoxy Titanium Chlorides

S/020/60/134/003/013/020 B016/B054

ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR (Institute of Elemental-organic Compounds of the

Academy of Sciences, USSR)

SUBMITTED:

June 9, 1960

Card 3/3

CIA-RDP86-00513R000204930010-2" APPROVED FOR RELEASE: 06/08/2000

2209, 1164, 1282

23589 s/062/61/000/005/004/009 B118/B208

AUTHORS:

Nesmeyanov, A.N., Nogina, O. V., and Berlin, A. M.

TITLE:

Mono-nacyclopentadienyl derivatives of titanium

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye khimicheskikh nauk, no. 5, 1961, 804 - 807

TEXT: The authors studied the reaction of sodium cyclopentadienyl with diethoxy-titanium dichloride

 $2Cl_2Ti(OC_2H_5)_2 + 3C_5H_5Na \rightarrow C_5H_5Ti(OC_2H_5)_3 + (C_5H_5)_2 Ti$ 

The separated liquid cyclopentadienyl-triethoxy-titanium which was analytically confirmed is not associated in solution, contrary to the common alkoxy-titanium derivatives. Its molecular weight, cryoscopically determined in benzene, corresponds to that of the monomer compound, even in a concentration of 1.7 mole %. Its structure was confirmed by conversion with acetyl chloride to cyclopentadienyl titanium trichloride C5H5TiCl3. Card 7/4

Mono-m-cyclopentadienyl ...

S/062/61/000/005/004/009 B118/B208

Its melting point is by 20°C higher than that found by R. D. Gorsich (see below). The second yellow, crystalline product of the above reaction corresponds to dicyclopentadienyl titanium ethoxy-chloride. To obtain only cyclopentadienyl trialkoxy-titanium in the reaction of sodium cyclopentadienyl with alkoxy titanium chlorides, tri-n-propoxy-titanium ohloride was used as titanium derivative, which gave, as was expected, cyclopentadienyl-tripropoxy-titanium  ${}^{C}_{5}H_{5}Ti(0C_{3}H_{7})_{3}$  in a yield of 96%. With acetyl chloride the latter gives cyclopentadienyl titanium trichloride:

 $c_5H_5Ti(oc_3H_7)_3 + 3cH_3cocl \rightarrow c_5H_5Ticl_3 + 3cH_3cocc_3H_7$ 

The reaction of sodium cyclopentadienyl with ethoxy-titanium trichloride took place according to the formula

 $c_{13}^{\text{TiOC}_2H_5} + 2c_{5}^{\text{H}_5}^{\text{Na}} - (c_{5}^{\text{H}_5})^{\text{Ti}} c_{2}^{\text{C1}} + NaC1$ 

The attempt of an ester interchange of cyclopentadienyl triethoxy-titanium with propyl alcohol was unsuccessful, as this alcohol converts this

Card 2/4

Mono- $\pi$ -cyclopentadienyl...

23589 \$/062/61/000/005/004/009 B118/B208

π-cyclopentadienyl compound to tetrapropoxy-titanium (C3H70)4Ti under mild conditions. Ethyl alcohol reacts similarly forming tetraethoxytitanium (95 % yield) and cyclopentadiene (97 % yield, in the form of thallium cyclopentadienyl). To obtain mixed chloride alcoholates of  $\pi$ oyolopentadienyl titanium,  $C_5H_5Ti(OR)Cl_2$  and  $C_5H_5Ti(OR)_2Cl$ ,  $\pi$ -cyclopentadienyl propoxy-titanium was allowed to react with acetyl chloride ( 1:2 and 1:1), where  $\mathbf{c}_{5}\mathbf{H}_{5}\mathbf{Ti}(\mathbf{0}\mathbf{c}_{3}\mathbf{H}_{7})\mathbf{cl}_{2}$  and  $\mathbf{c}_{5}\mathbf{H}_{5}\mathbf{Ti}(\mathbf{0}\mathbf{c}_{3}\mathbf{H}_{7})_{2}\mathbf{cl}$ , respectively, resulted. The reaction products are green-yellow viscous liquids, not stable to atmospheric moisture, but stable when stored at 1 - 5°C. There are 11 references; 3 Soviet-bloc and 8 non-Soviet-bloc. The 4 references to English-language publications read as follows: C.L. Sloan, W. A. Barber, J. Amer. Chem. Soc. 81. 1364 (1959); M. A. Lynch, I. C. Brantley, Chem. Abstr. 52, 11126 (1958); A. K. Fischer, G. Wilkinson, J. Inorgan. Nuclear Chem. 2, 149 (1956); R. D. Gorsich, J. Amer. Chem. Soc. 80, 4744 (1958). ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR (Institute of Elemental Organic Compounds of the Academy of Sciences USSR)

Card 3/4

NESMEYANOV, A.N.; NOGINA, O.V.; BERLIN, A.M.; GIRSHOVICH, A.S.; SHATALOV, G.V.

Acyl and alkoxyl derivatives of bis-(cyclopentadienyl)titanium and the refraction increment of the -C5H5Ti group. Izv. AN SSSR Otd.khim.nauk no.12:2146-2151 D '61. (MIRA 14:11)

1. Institut elementoorganicheskikh soyedineniy Akademii nauk SSSR. (Titanium compounds)

KORSHAK, V.V.; KRONGAUZ, Ye.S.; BERLIN, A.M.

Organophosphorus polymers with P - N bonds. Izv.an SSSR.Otd. khim.nauk no.8:1412-1416 ag '62. (MIRA 15:8)

1. Institut elementoorganicheskikh soyedineniy AN SSSR. (Phosphorus organic compounds) (Polymers)

BERLIN, A.M.

AID Nr. 982-10 4 June

SYNTHESIS OF POLYPYRAZOLES (USSR)

Korshak, V. V., Ye. S. Krongauz, A. M. Berlin, and P. N. Gribkova. IN: Akademiya nauk SSSR. Doklady, v. 149, no. 3, 21 Mar 1963, 602-605. S/020/63/149/003/020/028

Four polypyrazoles (I) with alternating pyrazole rings in the backbone, of the type

Card 1/5

SYNTHESIS OF POLYPYRAZOLES [Cont'd]

8/020/63/149/003/020/028

where

No.	R	R'	R <sup>n</sup>	m.p., *C	Reaction temperature and pressure, *C/mm Hg
1	C6H4-0-C6H4	CH 3	(CH <sub>2</sub> )4	210-220	200-210/1
5	C6H4 (CH2)2 C6H4	CH,	(CH <sub>2</sub> ),	218-225	200/10-4
3	(CH <sub>2</sub> ) <sub>8</sub>	C <sub>6</sub> H <sub>5</sub>	(CH <sub>2</sub> )4	100-120	225-235/1
4	C6H4 (CH2)2 C6H4	CH3	-(1)	260-280	250/1

have been synthesized for the first time by the reaction of bis(diketones) of the type R' COCH<sub>2</sub>CO-R-COCH<sub>2</sub>COR' (II) with dicarboxylic acid dihydrazides

Card 2/5

# SYNTHESIS OF POLYPYRAZOLES[Cont'd]

8/020/63/149/003/020/028

of the type NH<sub>2</sub>NHCO-R"-CONHNH<sub>2</sub> (III). Polymers I are formed as a result of pyrazole ring closure (polycyclization) which occurs in two steps as follows:

The first step is the formation of a polyhydrazone (IV) from an equimolar mixture of II and III in boiling absolute ethanol. Compounds IV are green powders soluble in common organic solvents and do not have a sharp melting point. The reduced viscosity of 0.5% IV in cresol was as high as 0.4. The second step of the reaction is the ring closure of IV to form I in quantitative Card 3/5

SYNTHESIS OF POLYPYRAZOLES [Cont'd]

8/020/63/149/003/020/028

yields when IV is heated for 3 to 5 hrs at its melting point in an N<sub>2</sub> atmosphere under reduced pressure. Polymers I are yellow powders of mol. wt. 9200, soluble in cresol, dimethylformamide, concentrated H<sub>2</sub>SO<sub>4</sub>, and formic acid. Upon ring closure the polymer chain of IV decreases in length, causing a drop of reduced viscosity in cresol from 0.4 to 0.1. It is noted that the synthesis of I can be achieved in one step by the reaction of bis(4-acetoacetylphenyl) ethane with adipic acid dihydrazides in boiling benzyl alcohol. The structures of I and IV were determined by elemental analysis, IR and UV spectroscopy, and analysis of their alkaline or acid hydrolysis products. In the UV spectra of I and IV obtained from sebacyldiacetophenone, a bathochromic shift of 40 mu was observed with respect to 4, 4'-bis[3-(5-methyl-N-acetylpyrazolyl)]diphenylethane and 4, 4'-bis(acetoacetyl)-diphenylethane acetylhydrazone. Prolonged

Card 4/5

#### SYNTHESIS OF POLYPRAZOLES [Cont'd]

8/020/63/149/003/020/028

treatment of polymers I with concentrated H<sub>2</sub>SO<sub>4</sub> yielded a mixture of unidentified sulfonated products. Basic hydrolysis of I or IV in an aqueous 25% KOH solution boiling for 12 hrs caused backbone degradation. Bis(diketones) of the structure

where R = -, or CH<sub>2</sub>, form polyhydrazones which could not be converted to the polypyrazoles. [NI]

Card 5/5

KORSHAK, V.V.; KRONGAUZ, Ye.S.; BERLIN, A.M.

New method for the production of polypyrazoles. Dokl. AN SSSR 152 no.5:1108-1110 0 '63. (MIRA 16:12)

1. Institut elementoorganicheskikh soyedineni; AN SSSR. 2. Chlenkorrespondent AN SSSR (for Korshak).

ACCESSION NR: AP4042875

5/0062/64/000/007/1281/1288

AUTHOR: Korshak, V. V.; Krongauz, Ye. S.; Berlin, A. M.; Gribkova, P. N.; Sheina, V. Ye.

TITLE: Synthesis of polymers for the polycyclization reaction. Communication 1. Polypyrazoles

SOURCE: AN SSSR. Izvestiya. Seriya khimicheskaya, no. 7, 1964, 1281-1288

TOPIC TAGS: polymer, heat resistant polymer, polyhydrazone, polypyrazole, bis-(6-diketone), dicarboxylic acid dihydrazide, polycyclization reaction, polypyrazole structure, polypyrazole property

ABSTRACT: Polymers containing pyrazole rings have been synthesized in an attempt to produce new polymeric materials with improved heat resistance and chemical stability. Polypyrazoles were synthesized from bis-(β-diketones) of the R'COCH<sub>2</sub>CO-R-COCH<sub>2</sub>COR' type and dihydrazides of dicarboxylic acids. This reaction, designated as polycyclication, proceeds in two steps: 1) formation of polyhydrazones by the reaction of the carbonyl oxygen of the ketone with the end amine

Cord 1/3

#### ACCESSION NR: AP4042875

group of the hydrazide, which is accompanied by separation of water, and 2) formation of polypyrazoles by separation of a water molecule and closing of the ring. Polyhydrazones are prepared by heating equimolar amounts of the initial materials in absolute ethanol for 10—36 hr. Polypyrazoles are formed by heating polyhydrazones at 200—250C in nitrogen at 1—2 mm Hg for 3—5 hr. Polypyrazoles are yellowish powders soluble in cresol, dimethylformamide, and concentrated sulfuric and formic acids. They melt with decomposition at 220—260C, and thus do not exhibit the expected heat resistance. A polypyrazole was synthesized in one step by reacting 4,4 bis(aceto-acetyl) diphenylethane with the dihydrazide of adipic acid in boiling benzyl alcohol. Attempts to synthesize polypyrazoles in melts failed. From a study of the properties and structure of the synthesized polypyrazoles it was concluded that changes in the structure of the polymer backbone with the aim of increasing its rigidity will increase the melting point of the polypyrazoles. Orig. art. has: 2 tablas.

ASSOCIATION: Institut elementoorganicheskikh soyedineniy Akademii a nauk SSSR (Institute of Organoelemental Compounds, Academy of Sciences SSSR)

Card 2/3

ACCESSION NR: APLIOLOUST

\$/0190/64/006/006/1078/1086

AUTHORS: Korshak, V. V.; Krongauz, Ye. S.; Berlin, A. M.

TITLE: Synthesis of polymers by the polycyclization reaction. 5. Polypyrazoles

SOURCE: Vy\*sokomolekulyarny\*ye soyedineniya, v. 6, no. 6, 1964, 1078-1086

TOPIC TAGS: polycyclization reaction, branched diketone, adipic acid dihydrazide, keto enol tautomerism, polypyrazole, polyhydrazone

ABSTRACT: This is a continuation of an earlier work by the authors and P. N. Gribkova (Dokl. AN SSSR,149,602,1953 [Abstracter's note: 196327) on the interaction of bis-( $\beta$ -diketones) with the dihydrazide of adipic acid (DAA). The present investigation differed from the previous one in that instead of linear diketones it involved branched diketones of the type

Cord 1/3

ACCESSION NR: AP4040487

where the R is either absent or represents CH<sub>2</sub>, CH<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>, CH<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>4</sub>CH<sub>2</sub>, or CH<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CH<sub>2</sub>. The synthesis of these monomers with DAA was conducted by heating equimolecular quantities of the reactants either in absolute ethanol or in a melt for periods up to 10 hours at 80-170C. The obtained polyhydrazones or polypyrazoles were analyzed and their melting point, viscosity (in cresol or sulfuric acid), and infrared spectra were recorded. It was found that the reaction of tetraacetyldiethylbenzol-, of 4,4'-bis-(2",2"-diacetoethyl)diphenyl-, and of 4,4'-bis-(2",

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Card 2/3